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organophosphorus by novel electrochemical oxidation using Mg/Fe layered double hydroxide anode. In this paper, Mg/Fe layered double hydroxide (LDH) was prepared by a co-precipitation method. The chemical and physical properties of the precipitated Mg/Fe LDH were characterized by X-ray diffraction, Fourier-transform infrared spectroscopy, thermal gravimetric analysis,

and scanning electron microscopy. The results showed that the precipitated Mg/Fe LDH was uniform in size and structure and had a lamellar structure with a thickness of around 5 nm. EDS analysis revealed the presence of both Mg and Fe, which implies that the Fe(OH)3 layer and hydroxide layer were both distributed on the Mg(OH)2 surface. The kinetic oxidation process of

diazinon (DZN) was carried out under identical conditions as its mineralization. It was found that although the operating voltage of the Mg/Fe LDH anode was at 2.0 V, the current density of the DZN oxidation could be 50 times higher than that at 2.4 V. The results of the mineralization kinetics suggested that the electrochemical oxidation of DZN could be enhanced by

the synergistic effect of the fast oxidative ability of Mg/Fe LDH and the radical trap effect of the hydroxide layer. The Mg/Fe LDH anode could be a potential 79a2804d6b

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